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what is an unsaturated solution in chemistry thoughtco Nov 22 2021 jul 30 2019 in a supersaturated solution there is more dissolved solute than in a saturated solution the solute can easily fall out of solution by crystallization or precipitation special conditions may be needed to supersaturate a solution it helps to heat a solution to increase solubility so more solute can be added

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tonicity wikipedia Jul 19 2021 in chemical biology tonicity is a measure of the effective osmotic pressure gradient the water potential of two solutions separated by a partially permeable cell membrane tonicity depends on the relative concentration of selective membrane impermeable solutes across a cell membrane which determine the direction and extent of osmotic flux it is commonly used when describing

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of a solution that is present in the greatest amount it is the substance in which the solute
is dissolved usually a solvent is a liquid however it can be a gas solid or supercritical
fluid

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section 4 5 response header section 6 2 entity header crlf section 7 1 crlf message body
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a description of the master contractor s proposed solution to accomplish the specified work
requirements proposed solution a more detailed

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marketing concept describing the integration of enterprise communication services such as
instant messaging chat presence information voice including ip telephony mobility features
including extension mobility and single number reach audio web video conferencing fixed
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